LOGANIN AND A NEW IRIDOID GLUCOSIDE FROM GENTIANA PYRENAICA

JULIAN GARCIA,* EMMANUEL MPONDO MPONDO,

Laboratoire de Pharmacognosie, UFR de Pharmacie, Université Joseph Fourier-Grenoble 1, Domaine de la Merci, F-38706 La Tronche Cedex, France

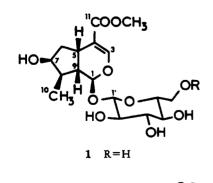
and ROBERT NARDIN

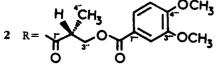
Centre Grenoblois de Résonance Magnétique, DRF LCH, CENG, Avenue des Martyrs, F-38041 Grenoble Cedex, France

ABSTRACT.—6'-[2(R)-methyl-3-veratroyloxy propanoyl] loganin [2], a new iridoid glucoside has been isolated from the aerial parts of *Gentiana pyrenaica* along with the known loganin [1]. Their structures have been established by spectral and chemical means.

Gentiana pyrenaica L. (Gentianaceae) was previously reported to contain six flavonoids (1) and two phthalide glucosides (2). Continuing the investigation for the chemical constituents of this species, the monoterpene components were studied. Herein we describe the isolation and characterization of 6'-[2(R)-methyl-3-veratroyloxy propanoyl] loganin [2], a new iridoid glucoside obtained from the aerial parts of this species along with the known iridoid loganin [1]. Their structures were determined by spectral and chemical evidence. This is the first time that iridoid glucosides are reported from G. pyrenaica.

Dried and powdered aerial parts of G. pyrenaica were extracted as described elsewhere (2). Chromatographic purifi-





cation of the CHCl₃ extract by centrifugal tlc, polyamide cc, and hplc afforded compounds 1 and 2.

Loganin [1] was identified by spectral and chromatographic comparison with an authentic sample and with literature values (3).

Compound 2 showed a positive vanillin reaction identical to that of 1, suggesting it to be an iridoid. In addition to the 240 nm absorption typical of a methoxycarbonyl enol ether system, the uv spectrum of 2 displayed two peaks at 255 and 290 nm indicating the presence of a phenolic acid unit in the molecule. Besides the typical loganin resonances, the ¹H-nmr spectrum exhibited signals of a 1,3,4-trisubstituted aromatic ring with two methoxyl groups characteristic of a veratroyl moiety (4). This one was in accordance with fabms data which showed fragments at m/z 165 (fab⁺) and m/z 181 (fab⁻). The ¹H-nmr spectrum, completed by spin decoupling experiments, displayed further signals corresponding to a β-hydroxyisobutyroyl unit (5). The chemical shift value of both the H-3"A and H-3"B (4.43 and 4.36 ppm) indicated that this unit was acylated at C-3" with veratric acid. This result was in agreement with fab⁺ms which showed two ions at m/z251 and 268 arising from the β -hydroxyisobutyric acid esterified with veratric acid. Evidence for location of the latter fragment at C-6' of the glucose

-

was given by the downfield shift of the C-6' methylene signals in the ¹H- and ¹³C-nmr spectra. As expected, alkaline hydrolysis of 2 afforded loganin, veratric acid, and β -hydroxyisobutyric acid; the latter had a negative optical rotation {[α]²⁵D-25° (c=0.105, MeOH)} indicating that it had the *R* configuration (6). From the above data compound 2 was characterized as 6'-[2(*R*)-methyl-3-veratroyloxy propanoyl] loganin, a new natural product.

In the family Gentianaceae, loganin derivatives were first isolated from Gentiana pedicellata (3,7) which like G. pyrenaica belongs to the Chondrophylla section. However, these iridoid glucosides differ from those of G. pyrenaica by the acylating unit and by its location on the glucose.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.— General isolation methods and instrumentation were similar to those used for the analysis of *G. pedicellata* (3). ¹H- and ¹³C-nmr spectra were obtained, respectively, on a Bruker AM 400 and AM 300 with TMS as internal standard. Fabms spectra were recorded on a Nermag R 10-10C.

ISOLATION.—Dried and powdered aerial parts of *G. pyrenaica* (240 g) were extracted as reported previously (2). CHCl₃ extract (7 g) was chromatographed by Si gel centrifugal tlc eluting by CHCl₃/MeOH with increasing MeOH content. Fractions eluted with CHCl₃-MeOH (9:1) were fractionated by polyamide cc [C₆H₆-MeOH (97:3)] yielding 2 (10 mg) after purification using RP-18 hplc [MeOH-H₂O (55:45)]. Compound 1 (2 mg) was obtained from the fraction eluted with CHCl₃-MeOH (6:4) and purified by RP-18 hplc [MeOH-H₂O (65:35)].

LOGANIN [1].—Identified by comparison of tlc, hplc, and 1 H- and 13 C-nmr spectra with authentic sample and literature values (3).

COMPOUND 2.—Colorless amorphous powder: $[\alpha]^{25}D - 32^{\circ}$ (c = 0.80, MeOH); uv λ max (MeOH) 225, 240, 255, 290; fab⁺ms [M + Na]⁺ 663, {M + H]⁺ 641, [M - aglycone]⁺ 413, 268, 151, 165; fab⁻ms [M - H]⁻ 639, 181, [M - glc acylated]⁻ 227; ¹H and ¹³C nmr see Table 1.

ALKALINE HYDROLYSIS OF 2.—Compound 2 (8 mg) was dissolved in 1 ml MeOH and 1 ml 1N NaOH. After 1 h at room temperature the reac-

TABLE 1.	¹ H- and ¹³ C-nmr data (400/75 MHz,
	CD_3OD) of compound 2 .

(D_3OD) or compound 2.			
Position	¹ H ^a	¹³ C	
1	5.06d(5)	98.3	
3	7.34d(1.5)	150.3	
4	_	113.9	
5	3.08 m	32.5	
6a	1.51 ddd (14,8.5,5)	42.9	
6Ь	2.22 ddd (14,7.5,1.5)	42.9	
7	3.99 m	74.9	
8	1.81 dqd (9,7,5)	42.4	
9	1.94 td (9,5)	46.5	
10	1.04d(7)	13.8	
11		169.5	
11-OMe	3.67 s	51.6	
1'	4.59 d (8)	100.3	
2'	3. 18 dd (9,8)	74.8	
3'	3.30–3.35 m	77.7	
4'	5.50 - 5.57 m	71.7	
5'	3.48 ddd (9,6.5,2)	75.8	
6'a	4.59 dd (12,2)	64.9	
б'Ь	4.18 dd (12,6.5)	04.9	
1″	—	175.2	
2″	2.99 quint. d (7,5.5)	40.8	
3"a	4.43 dd (10.5,7)	67.0	
3″Ь	4.36 dd (10.5,5.5)		
4″	1.23 d(7)	14.1	
Ar-CO	—	167.5	
1‴		123.5	
2‴	7.50d(2)	113.6	
3‴	→ ·	150.3	
4‴		155.1	
5‴	7.02 d (8)	112.2	
6‴	7.63 dd (8,2)	125.0	
ОМе	3.89-3.87 2s	56.6	

^aValues in parentheses are coupling constants in Hz.

tion mixture was neutralized with HCl and extracted with CHCl₃ and EtOAc. The CHCl₃ and EtOAC layers afforded, respectively, veratric acid and β -hydroxyisobutyric acid, both identified by comparison of their spectral data (¹H nmr) with authentic samples. The aqueous layer yielded loganin identified by comparison with a sample by tlc and hplc (in the above conditions loganic acid was not detected).

ACKNOWLEDGMENTS

We thank Mr. C. Bosso, CERMAV, Grenoble, for fabms measurements and Miss N. Durand for secretarial help.

LITERATURE CITED

1. A. Marston, K. Hostettmann, and A. Jacot-Guillarmod, Helv. Chim. Acta, 59, 2596 (1976). Mar-Apr 1989]

- J. Garcia, E. Mpondo Mpondo, A.J. Chulia, M. Kaouadji, and G. Cartier, *Phytochemistry*, in press.
- 3. J. Garcia and A.J. Chulia, *Planta Med.*, **52**, 327 (1986).
- K.S. Verma, G.R. Sood, S.R. Gupta, and V.K. Gujral, J. Chem. Soc., Perkin Trans. 1, 2473 (1979).
- 5. A.G. Gonzalez, J.M. Arteaga, and J.L. Breton, *Phytochemistry*, 14, 2039 (1975).
- 6. W. Choy, P. Ma, and S. Masamune, Tetrahedron Lett., 22, 3555 (1981).
- 7. J. Garcia and A.J. Chulia, *Planta Med.*, **53**, 101 (1987).

Received 26 September 1988